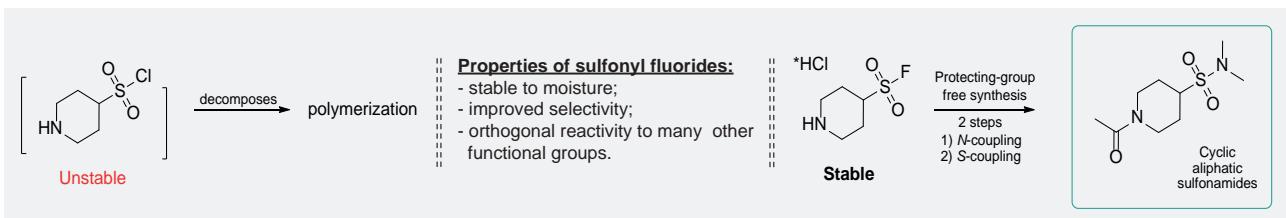


Aminosulfonyl fluorides for drug design

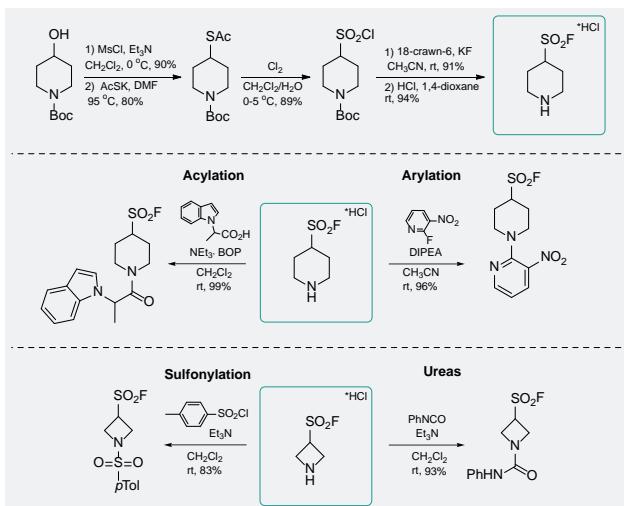
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Introduction and Aim

Sulfonyl chlorides ($-\text{SO}_2\text{Cl}$) are widely used in medicinal chemistry and agrochemistry as precursors to pharmacologically important sulfonamides. Many sulfonyl chlorides are unstable due to SO_2 extrusion and polymerization.¹ More stable sulfonyl fluorides ($-\text{SO}_2\text{F}$) in many cases are the only option to synthesize the desired sulfonamides. They are less reactive, so that they might even have free aliphatic amino groups in their structure and can be used for the protecting-group free synthesis of sulfonamides.² In the presence of the $-\text{SO}_2\text{F}$ group, the nitrogen atom can be modified by means of acylation, arylation, and reductive amination, that has a high potential for the synthesis of bioactive compounds. Herein, we aim to design and synthesize novel aminosulfonyl fluorides for drug discovery.



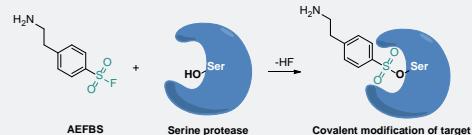
Synthesis



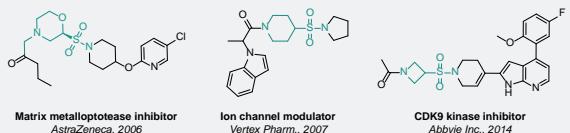
RSO_2F in Chemical Biology

► **RSO_2F reactive probes in chemical biology**
Over 40 approved drugs possess covalent mechanism of action³ and the interest for discovery of novel entities only tends to increase.⁴ High efficiency, selective binding with rare and specific targets, longer duration of action are among their main advantages. The $-\text{SO}_2\text{F}$ group covalently modifies many protein residues but in a context-specific manner: serine, threonine, tyrosine, lysine, cysteine, and histidine. Thus, sulfonyl fluorides can be uniquely enlisted in chemical probes, covalent inhibitors, or various stages of translational workflow.⁵

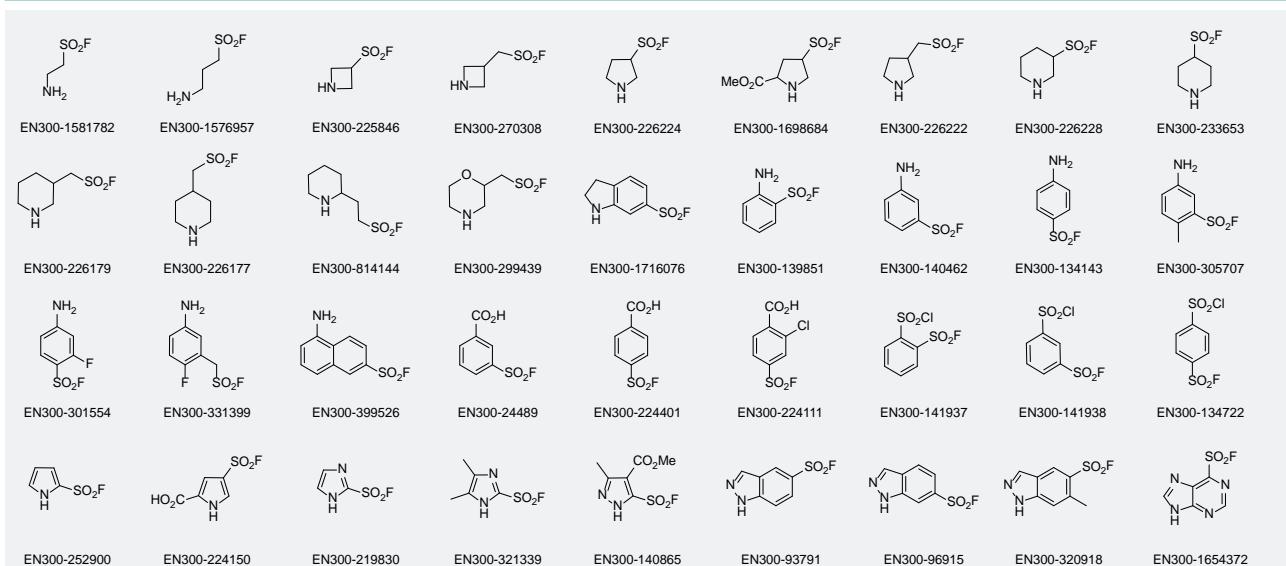
Mechanism of action



RSO_2NH_2 in drug discovery



Results



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